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catena-Poly[(*S*)-2-methylpiperazine-1,4-diium [[trichloridobismuthate(III)]-di-μ-chlorido]]

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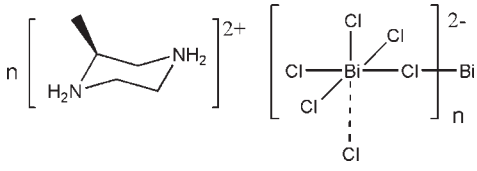
Key indicators: single-crystal X-ray study; T = 293 K; mean σ(C–C) = 0.011 Å; R factor = 0.031; wR factor = 0.066; data-to-parameter ratio = 26.2.

In the crystal structure of the title compound, $\{(C_5H_{14}N_2)[BiCl_5]\}_n$, the Bi^{III} cation is coordinated by six Cl^- anions in a distorted octahedral geometry. Two Cl^- anions bridge neighboring Bi^{III} cations, forming a zigzag polymeric chain along the *a* axis. The discrete methylpiperazinediium cation adopts a normal chair conformation and is linked to the polymeric chains by N—H···Cl hydrogen bonding.

Related literature

For transition-metal complexes of 2-methylpiperazine, see: Ye *et al.* (2009).



Experimental

Crystal data

$(C_5H_{14}N_2)[BiCl_5]$
M_r = 488.41
Orthorhombic, $P2_12_12_1$
a = 7.719 (1) Å
b = 10.8997 (16) Å
c = 16.302 (3) Å
V = 1371.6 (3) Å^3
Z = 4
Mo $K\alpha$ radiation
μ = 13.79 mm^{-1}
T = 293 K
0.28 × 0.26 × 0.24 mm

Data collection

Rigaku SCXmini diffractometer
Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2005)
T_{min} = 0.8, T_{max} = 0.9
14082 measured reflections
3150 independent reflections
3009 reflections with $I > 2\sigma(I)$
R_{int} = 0.089

Refinement

$R[F^2 > 2\sigma(F^2)]$ = 0.031
$wR(F^2)$ = 0.066
S = 1.03
3150 reflections
120 parameters
H-atom parameters constrained
$\Delta\rho_{max}$ = 1.57 e Å^{-3}
$\Delta\rho_{min}$ = −1.63 e Å^{-3}
Absolute structure: Flack (1983), 1327 Friedel pairs
Flack parameter: −0.021 (9)

Table 1
Selected bond lengths (Å).

Bi1—Cl1	2.8245 (18)	Bi1—Cl4	2.6135 (18)
Bi1—Cl2	2.597 (2)	Bi1—Cl5	2.875 (2)
Bi1—Cl3	2.561 (2)	Bi1—Cl5^{i}	2.820 (2)

Symmetry code: (i) $x+\frac{1}{2}, -y+\frac{3}{2}, -z+2$.

Table 2
Hydrogen-bond geometry (Å, °).

D—H···A	D—H	H···A	D···A	D—H···A
N1—H6A···Cl1^{ii}	0.97	2.30	3.262 (7)	171
N1—H6B···Cl2	0.97	2.48	3.255 (7)	137
N1—H6B···Cl3	0.97	2.61	3.244 (6)	124
N2—H7A···Cl4^{iii}	0.97	2.33	3.242 (7)	156
N2—H7B···Cl1^{iv}	0.97	2.25	3.184 (6)	161

Symmetry codes: (ii) $-x+1, y-\frac{1}{2}, -z+\frac{3}{2}$; (iii) $x-\frac{1}{2}, -y+\frac{1}{2}, -z+2$; (iv) $x, y-1, z$.

Data collection: *CrystalClear* (Rigaku, 2005); cell refinement: *CrystalClear*; data reduction: *CrystalClear*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

This work was supported by a start-up grant from Anyang Institute of Technology, China.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: XU5013).

References

Flack, H. D. (1983). *Acta Cryst.* A**39**, 876–881.
Rigaku (2005). *CrystalClear*. Rigaku Corporation, Tokyo, Japan.
Sheldrick, G. M. (2008). *Acta Cryst.* A**64**, 112–122.
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supplementary materials

Acta Cryst. (2010). E**66**, m1079 [**doi:10.1107/S1600536810031417**]

catena-Poly[(*S*)-2-methylpiperazine-1,4-diium [[trichloridobismuthate(III)]-di-μ-chlorido]]

Z.-L. Ru

Comment

The chiral 2-methylpiperazine has shown tremendous scope in the synthesis of transition metal complexes (Ye *et al.*, 2009). The construction of new members of this family of ligands is an important direction in the development of coordination chemistry. we report here the crystal structure of the title compound.

In the crystal of the title compound, $C_5H_{14}N_2.BiCl_5$ (Fig.1), the Bi^{3+} cations are coordinated by six Cl^- anions with distances ranging from 2.561 (2) to 2.875 (2) Å (Table 1). The values of bond angles Cl–Bi–Cl are near to 90 or 180°, which make the $[BiCl_6]^{3-}$ octahedral geometry. The protonated piperazine ring adopts a chair conformation. The Bi^{3+} cations conneted through bridging chlorine atom to form a one-dimensional chain structure. The crystal structure is stabilized by intermolecular N—H···Cl hydrogen bonds (Table 2).

Experimental

A mixture of (*S*)-2-methylpiperazine (2 mmol, 0.2 g), $BiCl_3$ (2 mmol, 0.62 g) and 20% aqueous HCl (20 ml) in 10 ml water was heated at 353 K for 0.5 h. The reaction mixture was cooled slowly to room temperature, crystals of the title compound were formed after 8 d.

Refinement

All H atoms were placed in calculated positions, with C—H = 0.96 or 0.98 Å and N—H = 0.97 Å, and refined using a riding model, with U_{iso}(H) = 1.5U_{eq}(C) for methyl H atoms and 1.2U_{eq}(C,*N*) for the others.

Figures

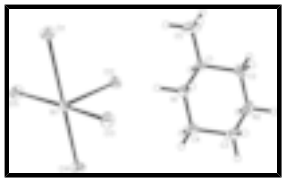

Fig. 1. The asymmetric unit of the title compound with atom labels. Displacement ellipsoids were drawn at the 30% probability level.

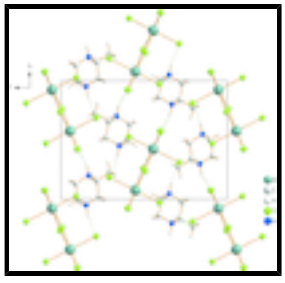

Fig. 2. The packing viewed along the *a* axis. Hydrogen bonds are drawn as dashed lines

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Crystal data

$(C_5H_{14}N_2)[BiCl_5]$	$F(000) = 904$
$M_r = 488.41$	$D_x = 2.365$ Mg m^{-3}
Orthorhombic, $P2_12_12_1$	Mo $K\alpha$ radiation, $\lambda = 0.71073$ Å
Hall symbol: P 2ac 2ab	Cell parameters from 3009 reflections
$a = 7.719$ (1) Å	$\theta = 2.5$–$27.5°$
$b = 10.8997$ (16) Å	$\mu = 13.79$ mm^{-1}
$c = 16.302$ (3) Å	$T = 293$ K
$V = 1371.6$ (3) Å^3	Block, colorless
$Z = 4$	0.28 × 0.26 × 0.24 mm

Data collection

Rigaku SCXmini diffractometer	3150 independent reflections
Radiation source: fine-focus sealed tube	3009 reflections with $I > 2\sigma(I)$
graphite	$R_{int} = 0.089$
Detector resolution: 13.6612 pixels mm^{-1}	$\theta_{max} = 27.5°$, $\theta_{min} = 2.5°$
ω scans	$h = -9 \rightarrow 10$
Absorption correction: multi-scan (*CrystalClear*; Rigaku, 2005)	$k = -14 \rightarrow 14$
$T_{min} = 0.8$, $T_{max} = 0.9$	$l = -21 \rightarrow 21$
14082 measured reflections	

Refinement

Refinement on F^2	Hydrogen site location: inferred from neighbouring sites
Least-squares matrix: full	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.031$	$w = 1/[\sigma^2(F_o^2) + (0.P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.066$	$(\Delta/\sigma)_{max} = 0.001$
$S = 1.03$	$\Delta\rho_{max} = 1.57$ e Å^{-3}
3150 reflections	$\Delta\rho_{min} = -1.63$ e Å^{-3}
120 parameters	Extinction correction: *SHELXL97* (Sheldrick, 2008), $Fc^* = kFc[1+0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$
0 restraints	Extinction coefficient: 0.0300 (5)
Primary atom site location: structure-invariant direct methods	Absolute structure: Flack (1983), 1327 Friedel pairs
Secondary atom site location: difference Fourier map	Flack parameter: −0.021 (9)

Special details

Geometry. All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2, conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2. The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc*. and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (Å^2)

	x	*y*	*z*	U_{iso}*/U_{eq}
Bi1	0.33206 (3)	0.57633 (2)	0.946205 (14)	0.02412 (11)
C1	0.1786 (10)	0.1571 (6)	0.7856 (4)	0.0296 (15)
H1	0.0737	0.1733	0.8177	0.036*
C2	0.1931 (12)	0.0212 (7)	0.7715 (5)	0.0391 (19)
H2A	0.0899	−0.0077	0.7436	0.047*
H2B	0.2917	0.0049	0.7362	0.047*
C3	0.3672 (10)	−0.0026 (8)	0.8968 (6)	0.049 (2)
H3A	0.4720	−0.0201	0.8661	0.058*
H3B	0.3740	−0.0456	0.9487	0.058*
C4	0.3548 (12)	0.1343 (8)	0.9124 (4)	0.041 (2)
H4A	0.2566	0.1512	0.9477	0.050*
H4B	0.4589	0.1625	0.9399	0.050*
C5	0.1685 (13)	0.2283 (9)	0.7064 (5)	0.059 (2)
H5A	0.1539	0.3139	0.7183	0.089*
H5B	0.0717	0.1997	0.6748	0.089*
H5C	0.2734	0.2166	0.6758	0.089*
Cl1	0.3347 (3)	0.68787 (17)	0.78982 (11)	0.0414 (4)
Cl2	0.0920 (2)	0.4253 (2)	0.89851 (12)	0.0378 (4)
Cl3	0.5685 (2)	0.4239 (2)	0.90419 (13)	0.0389 (4)
Cl4	0.3272 (3)	0.47345 (19)	1.09100 (11)	0.0412 (4)
Cl5	0.0888 (3)	0.7630 (2)	0.99374 (14)	0.0414 (5)
N1	0.3338 (8)	0.2005 (5)	0.8340 (4)	0.0331 (13)
H6A	0.4376	0.1895	0.8012	0.040*
H6B	0.3211	0.2875	0.8449	0.040*
N2	0.2143 (8)	−0.0475 (5)	0.8499 (4)	0.0387 (16)
H7A	0.1108	−0.0363	0.8828	0.046*
H7B	0.2270	−0.1345	0.8388	0.046*

Atomic displacement parameters (Å^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
Bi1	0.02674 (15)	0.02120 (14)	0.02443 (15)	−0.00003 (11)	−0.00047 (11)	−0.00161 (10)

C1	0.033 (3)	0.030 (4)	0.026 (3)	0.001 (3)	0.002 (3)	0.005 (3)
C2	0.054 (5)	0.032 (4)	0.031 (4)	−0.013 (4)	−0.001 (4)	−0.007 (3)
C3	0.045 (5)	0.040 (5)	0.060 (5)	−0.004 (4)	−0.017 (4)	0.021 (4)
C4	0.051 (5)	0.045 (5)	0.028 (4)	−0.013 (4)	−0.011 (4)	0.011 (3)
C5	0.062 (5)	0.076 (7)	0.040 (5)	0.004 (7)	0.000 (5)	0.023 (5)
Cl1	0.0513 (10)	0.0414 (10)	0.0316 (9)	0.0075 (11)	0.0079 (10)	0.0036 (8)
Cl2	0.0344 (9)	0.0359 (10)	0.0432 (10)	−0.0050 (9)	−0.0056 (8)	−0.0058 (10)
Cl3	0.0326 (8)	0.0350 (10)	0.0490 (11)	0.0048 (9)	0.0058 (8)	−0.0045 (11)
Cl4	0.0382 (9)	0.0521 (11)	0.0333 (9)	−0.0029 (11)	−0.0032 (9)	0.0123 (8)
Cl5	0.0440 (9)	0.0356 (11)	0.0446 (11)	0.0132 (8)	0.0096 (8)	0.0003 (9)
N1	0.043 (3)	0.026 (3)	0.031 (3)	−0.005 (3)	0.000 (3)	0.002 (2)
N2	0.039 (3)	0.024 (3)	0.053 (4)	−0.002 (3)	−0.003 (3)	0.001 (3)

Geometric parameters (Å, °)

Bi1—Cl1	2.8245 (18)	C3—C4	1.517 (12)
Bi1—Cl2	2.597 (2)	C3—H3A	0.9700
Bi1—Cl3	2.561 (2)	C3—H3B	0.9700
Bi1—Cl4	2.6135 (18)	C4—N1	1.476 (9)
Bi1—Cl5	2.875 (2)	C4—H4A	0.9700
Bi1—Cl5^{i}	2.820 (2)	C4—H4B	0.9700
C1—C2	1.504 (10)	C5—H5A	0.9600
C1—C5	1.509 (10)	C5—H5B	0.9600
C1—N1	1.510 (10)	C5—H5C	0.9600
C1—H1	0.9800	N1—H6A	0.9700
C2—N2	1.490 (10)	N1—H6B	0.9700
C2—H2A	0.9700	N2—H7A	0.9700
C2—H2B	0.9700	N2—H7B	0.9700
C3—N2	1.489 (10)		
Cl3—Bi1—Cl2	90.97 (6)	C4—C3—H3A	109.4
Cl3—Bi1—Cl4	88.48 (7)	N2—C3—H3B	109.4
Cl2—Bi1—Cl4	89.32 (7)	C4—C3—H3B	109.4
Cl3—Bi1—Cl5^{i}	89.71 (8)	H3A—C3—H3B	108.0
Cl2—Bi1—Cl5^{i}	177.10 (7)	N1—C4—C3	110.0 (6)
Cl4—Bi1—Cl5^{i}	87.88 (7)	N1—C4—H4A	109.7
Cl3—Bi1—Cl1	91.88 (7)	C3—C4—H4A	109.7
Cl2—Bi1—Cl1	90.44 (7)	N1—C4—H4B	109.7
Cl4—Bi1—Cl1	179.57 (7)	C3—C4—H4B	109.7
Cl5^{i}—Bi1—Cl1	92.35 (7)	H4A—C4—H4B	108.2
Cl3—Bi1—Cl5	175.20 (8)	C1—C5—H5A	109.5
Cl2—Bi1—Cl5	93.63 (8)	C1—C5—H5B	109.5
Cl4—Bi1—Cl5	92.91 (7)	H5A—C5—H5B	109.5
Cl5^{i}—Bi1—Cl5	85.753 (13)	C1—C5—H5C	109.5
Cl1—Bi1—Cl5	86.75 (6)	H5A—C5—H5C	109.5
C2—C1—C5	112.3 (7)	H5B—C5—H5C	109.5
C2—C1—N1	109.2 (7)	Bi1^{ii}—Cl5—Bi1	172.74 (9)
C5—C1—N1	109.0 (6)	C4—N1—C1	112.7 (6)

C2—C1—H1	108.7	C4—N1—H6A	109.0
C5—C1—H1	108.7	C1—N1—H6A	109.2
N1—C1—H1	108.7	C4—N1—H6B	109.3
N2—C2—C1	111.8 (6)	C1—N1—H6B	108.8
N2—C2—H2A	109.3	H6A—N1—H6B	107.8
C1—C2—H2A	109.3	C3—N2—C2	111.2 (6)
N2—C2—H2B	109.3	C3—N2—H7A	109.1
C1—C2—H2B	109.3	C2—N2—H7A	108.7
H2A—C2—H2B	107.9	C3—N2—H7B	109.7
N2—C3—C4	111.1 (7)	C2—N2—H7B	110.0
N2—C3—H3A	109.4	H7A—N2—H7B	108.0

Symmetry codes: (i) *x*+1/2, −*y*+3/2, −*z*+2; (ii) *x*−1/2, −*y*+3/2, −*z*+2.

Hydrogen-bond geometry (Å, °)

D—H···*A*	*D*—H	H···*A*	*D*···*A*	*D*—H···*A*
N1—H6A···Cl1^{iii}	0.97	2.30	3.262 (7)	171
N1—H6B···Cl2	0.97	2.48	3.255 (7)	137
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N2—H7A···Cl4^{iv}	0.97	2.33	3.242 (7)	156
N2—H7B···Cl1^{v}	0.97	2.25	3.184 (6)	161

Symmetry codes: (iii) −*x*+1, *y*−1/2, −*z*+3/2; (iv) *x*−1/2, −*y*+1/2, −*z*+2; (v) *x*, *y*−1, *z*.

Fig. 1

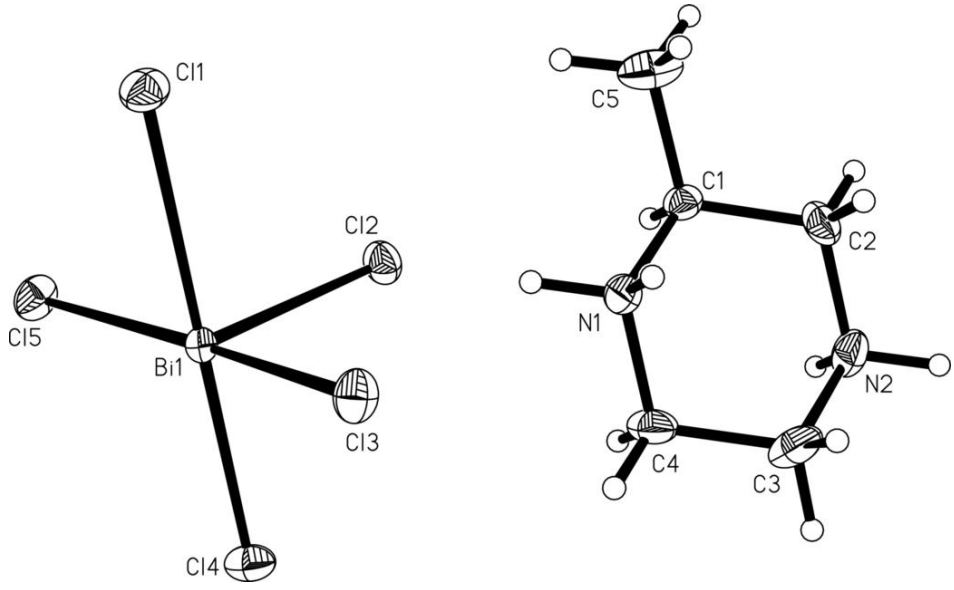

Fig. 2